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RF (1989) Redundancies in protein sequences In Prediction of Protein Structure and the Principles of Protein Conformation (Fasman, GD, ed), pp 599-623, Plenum Press, New York Cysteine Asparagine Glutamine Positively charged R groups Lysine Histidine Arginine Cys Asn Gin Lys His Arg c N Q K 121 132 146 146 155 174 196 202 217 218 182

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D) is equally useful with solutions of acetic acid and of hydrochloric acid E) relates the pH of a solution to the pKa and the concentrations of acid and conjugate base 17 Consider an acetate buffer, initially at the same pH as its pKa (476) When sodium hydroxide (NaOH) is ...

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a Which of the following aqueous solutions has the lowest pH: 01 M HCl; 01 M acetic acid (pK a 486); 01 M formic acid (pK a 375)? Answer A 01 M HCl solution has the lowest pH because HCl is a strong acid and dissociates completely to H Cl , yielding the highest [H] 9 Meanings of K a and pK

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IUBMB – International Union of Biochemistry and Molecular Biology Major class sub-sub class more information Reaction – Substrate converted into product Rate constant with forward and reverse reaction Rate equation – Reaction velocity Reaction order – determined experimentally vary one [reactant], hold the rest constant

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